



# Ferric Nitrate Solution, 30%

#### Section 1 - Identification

Ferric Nitrate Solution, 30% Product

Recommended Use:

Utilized in industrial chemical applications.

TradeMark Nitrogen Corp. Manufacturer

Address 1216 Old Hopewell Road, Tampa, FL 33619

Phone (813) 626-1181 (800) 452-3107

24 Hour Chemtrec **Emergency** (800) 424-9300

Contact

#### Section 2 - Hazard Identification



P273

# Signal Word: DANGER

**Precautionary Statements:** P210 Keep away from heat/sparks/open flames/hot surfaces. - No smoking. P262 Do not get in eyes, on skin, or on clothing

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P281 Use personal protective equipment as required.

Avoid release to the environment.

P284 Wear respiratory protection.

P301 IF SWALLOWED: P331 Do NOT induce vomiting. P313 Get medical advice/attention. P303 IF ON SKIN OR HAIR:

Remove/Take off immediately all contaminated clothing. Rinse P361 P353

skin with water/shower.

IF INHALED P304

P340 Remove victim to fresh air and keep at rest in a position

comfortable for breathing.

P313 Get medical advice/attention.

P305 IF IN FYFS

P351 P338 Rinse cautiously with water for several minutes. Remove contact

lenses, if present and easy to do. Continue rinsing.

P337 P313 If eye irritation persists: Get medical advice/attention.

P370 P378 In case of fire: Use water for extinction.

P410 P403 Protect from sunlight. Store in a well-ventilated place.

P402 Store in a cool, dry place.

#### Section 3 - Composition

Ingredients	Component	CAS. No.	Percent by Weight	Percent as Meta
	Ferric (III) Nitrate (Fe(NO <sub>3</sub> ) <sub>3</sub> )	10421-48-4	30.3%	7.0%
	Nitric Acid (HNO <sub>3</sub> )	7697-37-2	0-0.1%	
	Water (H <sub>2</sub> 0)	7732-18-5	Balance	

**Hazard Statements:** 

H303 May be harmful if swallowed

H314 Causes severe skin burns and eye damage

H335 May cause respiratory irritation H373

May cause damage to organs through

prolonged or repeated exposure

Section 4 - Fire	st Aid Measures
Inhalation	If inhaled: Remove person to fresh air and keep comfortable for breathing. Provide artificial respiration if necessary. Seek medical attention if necessary.
Skin Contact	If on skin (or hair): Take off all contaminated clothing. Rinse skin with soap and water for at least 15 minutes.
Eye Contact	If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing for at least 15 minutes. Seek medical attention if irritation persists.
Ingestion	If swallowed: <b>Do NOT induce vomiting</b> . Drink large amounts of water. Never give anything by mouth to an unconscious person. Seek medical attention.
Acute Health Hazards	Harmful if swallowed or inhaled. Destructive to muccous membranes and upper respiratory tract, eyes and skin. Redness and irritation of tissue may occur.
Chronic Health Hazards	Excess iron intake can lead to cell damage, lipid peroxidation and DNA mutagenesis. In severe cases it can lead to hemochromatosis. Other chronic effects can include metabolic syndrome, type 2 diabetes, sarcopenia, non-alcoholic fatty liver disease and Alzhimer's and other neurodegenerative diseases.

Section 5 - Fire	Fighting Measures
Suitable	Non-combustible, but can o

contribute to the intensity of the fire. Wear self-contained breathing apparatus and full protective gear.

Extinguishing Use water spray - not water jet

Techniques & Equipment Chemical

This material may decompose and produce acrid vapors and oxides of nitrogen and carbon.

Hazards From Fire

Special Fire Use water spray. CO<sub>2</sub> or halon may provide limited control.

Fighting Procedures NFPA Rating

Health - 1 (Slight) Fire - 0 (Least) Reactivity - 1 (Slight)

Other Do not allow run-off from fire fighting to enter drains or water courses.



## Section 6 - Accidental Release Measure

Avoid splashing. Prevent exposure to spilled material with the use of proper PPE. Personal Precautions

Protective PPE should include gloves, goggles, face shield and level C protective suit.

Permissible

Equipment

Control the flow of product using dikes of soil, sand bags or other commercially available inert sorbent socks or booms. Containment

In Case of Spill Absorb product with inert absorbent. Avoid splashing or spraying. Contain and pick up spill in diked area. Prevent discharge to sewers or water ways. If

uncontaminated, recover and re-use.

### Section 7 - Safe Handling & Storage

Safe Handling & labeled properly. Storage

Exposure Limits Component

Precautions for Recommended storage above 50°F. Store in a well ventilated cool, dry place out of direct sunlight. Prevent from freezing. Containers should be kept closed and

Immediately

Avoid contamination with combustible materials. Keep away from fire. Extreme heat may cause decomposing to toxic fumes of nitrogen oxides. Incompatibility

Threshold Limit Value Short Term Exposure

### Section 8 - Exposure Controls / Personal Protection

Dangerous to Life or Exposure Limit Limit Health Ferric (III) Nitrate Not Established Not Established Not Established Not Established  $(Fe(NO_3)_3)$ Nitric Acid (HNO<sub>3</sub>) 2 ppm (TWA) 25 ppm 2 ppm 4 ppm Water (H<sub>2</sub>O) Not Established Not Established Not Established Not Established Engineering

Provide ventilation sufficient to maintain exposure below PEL/TWA/TLV. Washing facilities should be available.

Controls

Personal Eyes Chemical safety goggles and full face shield Protective Hands Chemical resistant gloves with gaunlet.

Equipment

Respiratory None required under normal conditions. Self contained respiratory equipment should be used under spill situations.

Protective Clothing Chemical resistant protective clothing.









Gloves

Goggles

Face Shield

Protective Clothing

Section 9 - Physical & Chemical Properties Appearance and A reddish to brown liquid. Slight acrid Specific Gravity 1.1 - 1.4 Odor odor. **Boiling Point** > 212°F (>100°C) at 1 atmosphere Molecular Weight No Data Available Solubility in Water Freezing Point No Data Available No Data Available Vapor Pressure No Data Available Evaporative Rate No Data Available Weight per На 02-20 10.90 lbs./gal Gallon Salt-Out Temp < 55°F (10.0°C) Flash Point No Data Available Flammability No Data Available Auto Ignition Temp Not Flammable Limits UEL No Data Available LEL N/A

Section 10 -Stability & Reactivity

Reactivity Reacts with strong alkaline substances. This product may react with reducing agents. Stability Product is stable under normal conditions. Hazardous Hazardous polymerization will not occur. Reactions Conditions to Elevated temperatures may cause container to rupture. Avoid evaporation to dryness. Avoid Incompatible Avoid contact with cyanides, sulfides, sulfites, chlorine or chlorine bleaches, strong alkalis, mild steel, strong reducting agents and finely powdered metals.

Materials Hazardous Extreme heat may cause decomposing to toxic fumes of nitrogen oxides.

Decomposition Products

Section 11 - Toxicology Information

Routes of Inhalation, ingestion or skin/eye absorption Exposure Symptoms and Eyes May cause eye damage Signs of Skin Mild irritant.

Exposure Inhalation of gases or mist causes irritation to the upper respiratory system, including the mucous membranes of the nose, mouth and throat.

Coughing, fever, nausea, irritability, spasms, possible pneumonia, apathy, headaches, weakness and chemical burns if inhaled.

Inaestion may cause upset stomach.

Long Term Excess iron intake can lead to cell damage, lipid peroxidation and DNA mutagenesis. In severe cases it can lead to hemochromatosis. Other chronic effects can Effects include metabolic syndrome, type 2 diabetes, sarcopenia, nonalcoholic fatty liver disease and Alzheimer's and other neurodegenerative diseases.

The International Agency for Research on Cancer has not classified ferric nitrate for its carcinogenic potential (IARC 1987). Carcinogen

Section 12 **Ecological Information** 

No Data Available Water Section 13 -**Disposal Considerations** 

Waste

Disposal must be done in accordance with local, state and federal environmental regulations. Place waste in an appropriate container with correct labeling.

Section 14 - Transport Information

This material is hazardous as defined by 49 CFR 172.101 by the US Department of Transportation

**UN ID Number Proper Shipping** 

Name UN3093, Corrosive Liquid, Oxidizing, N.O.S. (Ferric Nitrate Solution), 8(5.1), II

Hazard Class Packing Group PG II US DOT Label Corrosive Emergency 154 Response Guide

Number This material is regulated as a Dangerous Good per the IMDG Code

**UN ID Number** UN3093

**Proper Shipping** 

UN3093, Corrosive Liquid, Oxidizing, N.O.S. (Ferric Nitrate Solution), 8(5.1), II Name Hazard Class 8(5.1)

Packing Group PG II Label Corrosive

This material is regulated as a Dangerous Good per the IATA Code

**UN ID Number** UN3093

**Proper Shipping** Name UN3093, Corrosive Liquid, Oxidizing, N.O.S. (Ferric Nitrate Solution), 8(5.1), II

Hazard Class 8(5.1) Packing Group PG II

851 = 1 Liter Or Less Packing 855 = More than 1 Instructions Liter (CARGO

AIRCRAFT ONLY)

Label Corrosive





Section 15 – I	Regulatory Inforr	nation					
United States -	This product has been reviewed according to the EPA Hazard Categories promulgated under Sections 311 and 312 of Title III of the Superfund Amendments and						
SARA Hazard	Reauthorization Act (SARA) and is considered, under applicable definitions, to meet the following categories:						
Category	Fire - No	Pressure - No	Reactive - Yes	Acute - Yes	Chronic - No		
SARA Title III Information	This product contains the following substances subject to the reporting requirements of Title III (EPCRA) of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR Part 372:						
	Chemical CAS No.	CAS No.	CERCLA RQ (lbs.)	SARA Reporting			
				302	304	313	
	Ferric Nitrate	10421-48-4	N/A	No	No	Yes	
	Nitric Acid	7697-37-2	1,000 lbs (453.6 Kg)	Yes	No	Yes	
CERCLA / Superfund, 40 CFR Part 117, 302		oduct contains components subject to substances designated as CERCLA reportable Quantity (RQ) Substances, it will be designated in the above table RQ value in pounds. If there is a release of RQ Substance to the environment, notification to the National Response Center, Washington DC (800-424-required.					
T004	N						

TSCA Nitric acid, iron(3+) salt (3:1) is on the Active TSCA inventory list.

California Prop

Components of this product are not listed on California's Prop 65 list.

# Section 16 - Other Information

Disclaimer

Issue Date	4/30/2024
Date of Revision	April 2024 Section 14 Transportation Information updated. September 2022 UN Number & proper shipping name updated based on DOT laboratory testing results.
	March 2021 document prepared in accordance with 29 CFR 1910.1200 Appendix D to meet Global Harmonization Standards.

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